



CFD Modeling of Hollow Fiber Membrane Contactor for Post-Combustion CO₂ Capture



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Outline

- Introduction
- Mass Transfer in a gas/liquid membrane contactor
- ☐ Development of CFD model
- ☐ Results and discussion
- Characterization of Liquid/gas film resistance
- Influence of membrane to overall mass transfer
- Influence of chemical reaction to overall mass transfer
- Conclusions





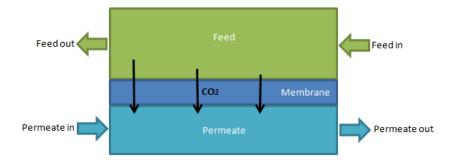
Introduction

Membrane contactor; a hybrid technology

□Compared to absorption columns

Larger sp. surface area, linear scale-up, modular design, flexible operation

Contractors	Sp surface area (m ² /m ³)	Reference
Free dispersion column	1-10	Reed et al. (1995)
Mechanically agitated column	50-150	Westerterp et al. (1984)
Packed column	100-800	Reed et al. (1995)
Membrane contactor	1500-3000	Kumar et al. (2002)





Amine Absorption Unit

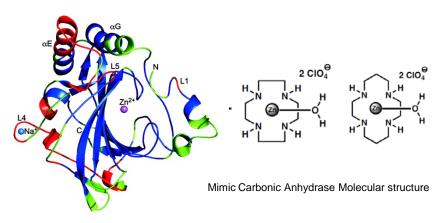




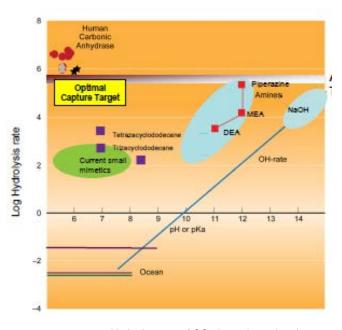
Introduction

■ Membrane contactor for flue gas treatment

- High and non dispersive contact offers efficient mass transfer
- Highly reactive Carbonic anhydrase (CA) and mimic CA with reaction rate
 4000 times faster than MEA to account for low mass transfer driving force



Carbonic Anhydrase Molecular structure



Hydration rate of CO2 by various absorbents.

Aines R. H Lawrence Livermore National Laboratory

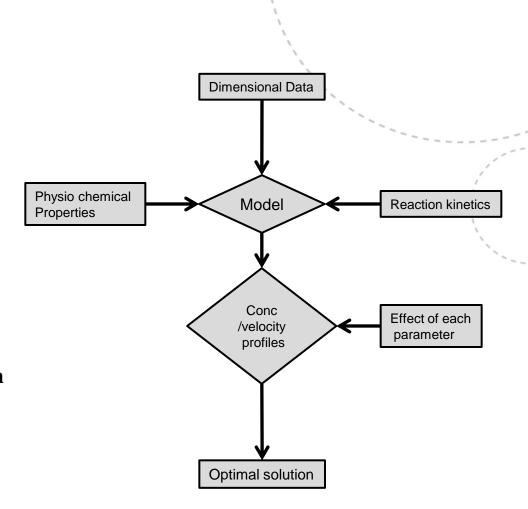


Innovation and Creativity

Scope

☐ This work

- CFD model of a membrane contactor with physical and chemical absorption by using Multi physics COMSOL
- Carbonic anhydrase (CA) and mimic CA for reactive absorption of CO2
- Mass transfer through a micro porous, hydrophobic membrane
- Developed model to suggest an optimal set of parameters for designing operations of a membrane contactor.

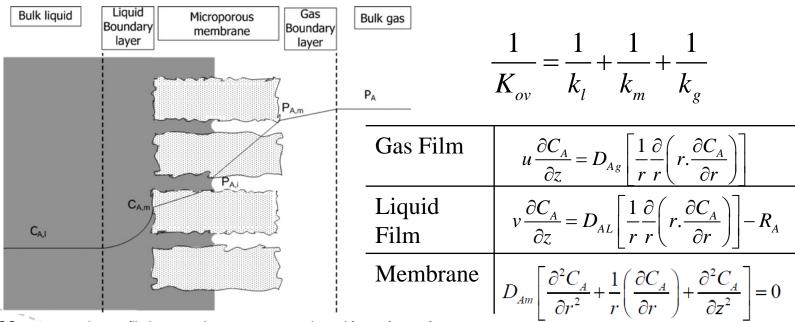






Mass Transfer in Membrane Contactor

□ Overall mass transfer in a membrane contactor is analogues to heat transfer and can be exemplified by resistance in series model.



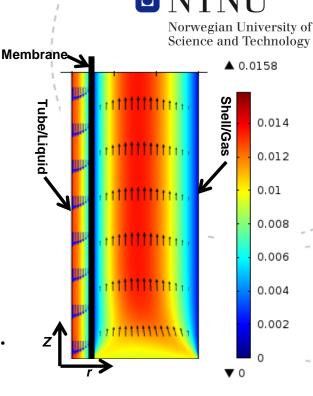
 ${\rm CO_2}$ concentration profile in a membrane contactor adapted from **Journal** of Membrane Science 380 (2011) 21–33



Model development

□ Assumptions

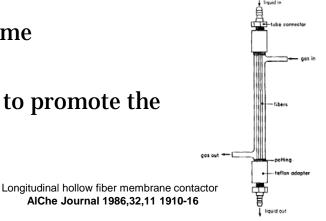
- Regular and uniform pores
- Hydrophobic membrane
- First order reaction between CO2 and catalyst.



Velocity profile in gas/ liquid phase

☐ Model Basis

- Velocities are adjusted to maintain a laminar regime
- Flue gas containing 10% CO2 at 1 bar and 25°C.
- Carbonic Anhydrase (CA) and mimic CA catalyst to promote the absorption in liquid



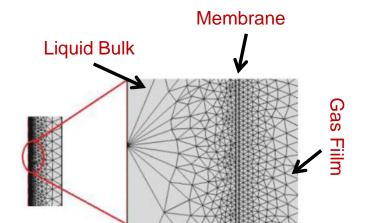




Model development (cont...)

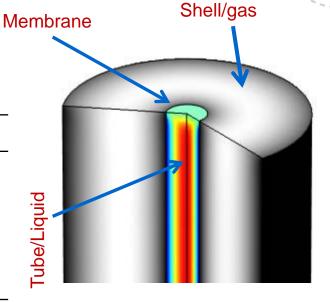
Basis for model geometry

■ In this work Finite Element Method is used for calculations.



Parameter	Symbol	Value
Length	L	100mm
Radius of membrane	R_1	1 mm
Radius of shell	R_2	6mm
Porosity	P_{or}	0.7
Tortuosity	T_{or}	2
Conc. in gas phase	C_{o}	10 %

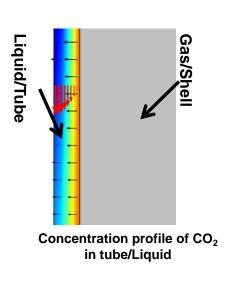
Literature	Value	Source
Diffusivity in Shell	$1.5 \times 10^{-5} \text{m}^2/\text{S}$	Reid, R. C.; Prausnitz, 1986
Diffusivity in water	$1.92 \times 10^{-9} \text{m}^2/\text{S}$	Chem. Eng. Comm., 1996, 144,113-158
Partition coefficient	1.205 x 10 ⁻²	J. Chern. Eng. Data, 1988, 33, 29-34,
Reaction rate of MEA	5920 (1/s)	Chem Eng. Sci. 2007,39,207-25

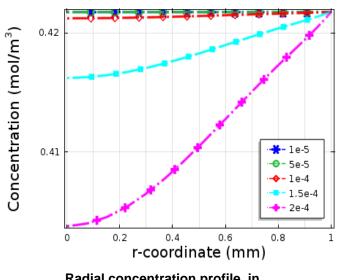


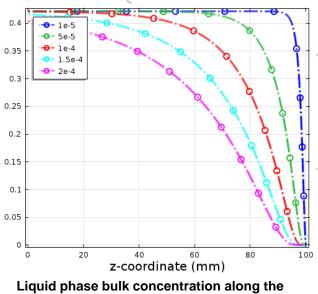




☐ Physical absorption of CO2 in liquid







Radial concentration profile in liquid phase

length of contactor.

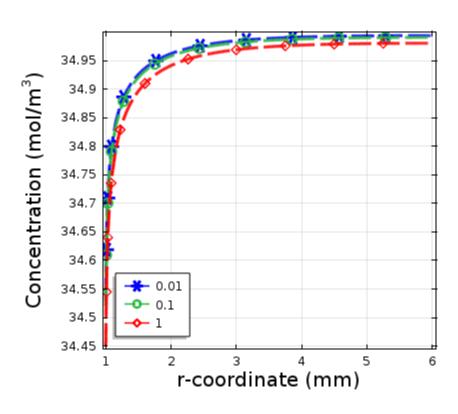
Liquid velocity changing from 1e-5 to 1 m/s

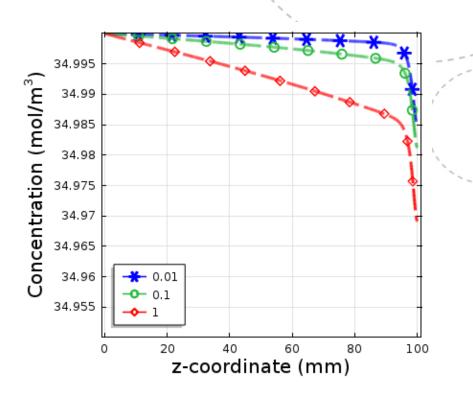
Appreciable effect on liquid loading.





☐ Physical absorption of CO2 in liquid



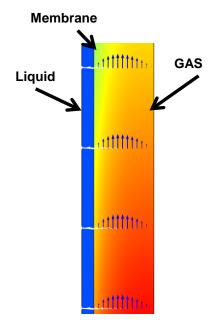


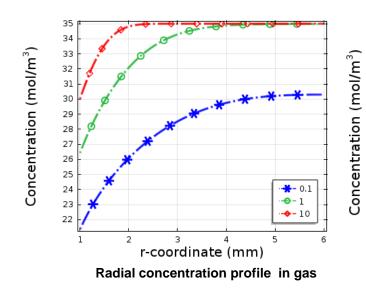
Appreciable effect on liquid loading.
 However, no significant effect on gas phase

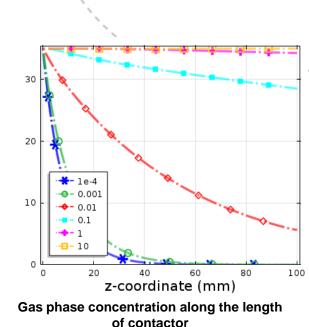




☐ Characterization of Gas film resistance







Concentration profile of CO₂ in shell and tube

Investigated gas velocity: 1e-4 to 10 m/s

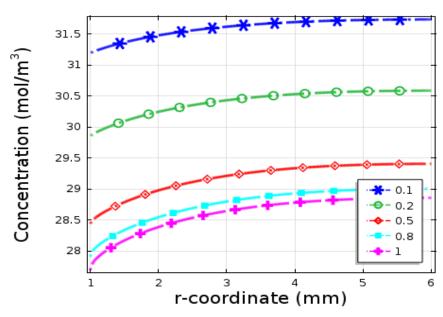
Appreciable effect of gas velocity at interfacial concentration shows that mass transfer limitation has now moved from liquid film to gas film/membrane.





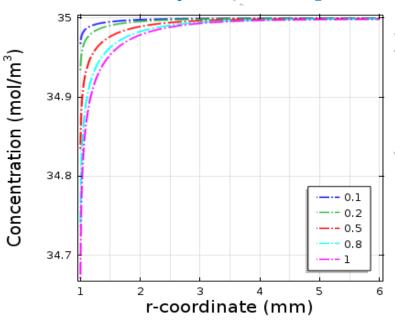
☐ Influence of porosity

Chemisorption



Gas phase radial concentration. porosity

Physical absorption



Gas phase radial concentration. porosity

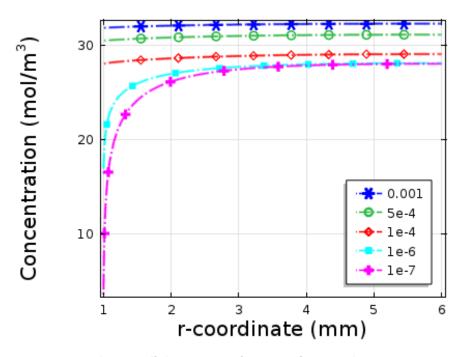
Porosity of membrane varied from 0.1 to 1.

Membrane resistance has significant contribution to mass transfer in chemical absorption .





☐ Influence of membrane thickness



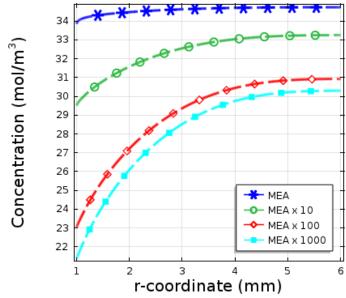
Gas phase radial concentration at varies membrane thickness

- Thickness of membrane was varied between 1 mm and 1e-4 mm.
- Thickness of membrane does have a significant effect on mass transfer but developing a self supported membrane of 1e-4 mm is challenging.

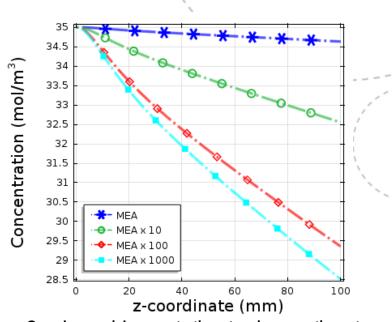




□ Influence of Reaction



Gas phase radial concentration at various reaction rates



Gas phase axial concentration at various reaction rates

Reaction rate varied 10, 100 and 1000 times that of MEA.

With increase in reaction rate, a significant increase in efficiency is observed





Conclusions

- For Physical absorption, liquid film resistance is the limiting factor to mass transfer.
- For Chemisorption mass transfer resistance shifts to membrane and gas film.
- Membrane porosity and thickness contributes significantly to mass transfer in chemical absorption.
- This work is a guideline to design a membrane contactor operating with and without chemical reaction.





Validation

Legends



Tank

Pump

Tube Bundle

vessel

Membrane Cell

> Gass Analyser

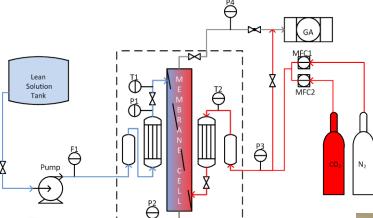
Solution Tank

Indicator

Cylinder

 \bowtie

Gate valve









Thank you for your attention!



Special thanks to:

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- Mr Hassan Ali